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Synthesis of bulky 1,2-dialkoxy- and 1,2,3-trialkoxy-arenes

Michel Stephan^{a,b,*,†}, Borut Zupančič^b, Barbara Mohar^{a,b,*}

^a En-Fist Centre of Excellence, Dunajska 156, SI-1000 Ljubljana, Slovenia
^b National Institute of Chemistry, Hajdrihova 19, SI-1001 Ljubljana, Slovenia

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ABSTRACT

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1. Introduction

1,2-Dimethoxy- and 1,2,3-trimethoxy-benzene units are present in a variety of organic molecules, which possess interesting chemical and biological properties.¹ For example, many representative members of the naturally occurring isoquinoline alkaloid and flavone families contain the first unit, and some cytotoxic lignans (e.g., podophyllotoxin, steganone), and alkaloids (e.g., reserpine) contain the second. A comprehensive literature survey revealed that the bulk of the starting syntheses of 1,2-di(linear alkoxy)arenes rely heavily on Williamson etherification of catechols or on an oxidation—hydrolysis sequence of 1-alkoxyarene-2-carboxaldehydes and subsequent etherification of the resulting aromatic OH group.² However, the few known sterically hindered analogues are prepared by inconvenient procedures or are poorly characterized, and the bulky trialkoxy-arenes are especially less well encountered in the literature and constitute a particular synthetic challenge.

As part of our continuous research focus on polyalkoxy-arylcontaining phosphine ligands for transition metal-catalyzed transformations,³ we present herein the practical synthesis on a preparative scale of series of bulky 1,2-dialkoxy-benzenes and naphthalenes, and also of isomeric trialkoxy-benzenes.

2. Results and discussion

A large series of bulky 1,2-dialkoxy- and 1,2,3-trialkoxy-benzenes was efficiently prepared via Wil-

liamson etherification. Preparation of their contiguous bromine-containing derivatives was also

2.1. Preparation of bulky 1,2-dialkoxy-arenes

A series of 1,2-dialkoxy-benzenes 1a-j was prepared under improved procedures starting from catechol and the appropriate alkyl halide via Williamson etherification in acetone, followed by hydrogenation of the pendent olefinic group when required (Scheme 1, Table 1).⁴ Various primary and secondary alkyl groups with branching on the first or second carbon atom were successfully introduced under mild conditions demonstrating the suitability of this method for the incorporation of the relatively sensitive β -methallyl (β -Met), 2-cyclohexenyl (*c*-Hexen-2-yl) and cyclohexyl (Cy) groups, which were appended taking advantage of the corresponding activated halide on allylic position toward nucleophilic substitution, thus avoiding harsh alkylation conditions.⁵ Following, PtO₂-catalyzed hydrogenation under low H₂ pressures of the olefinic groups proceeded smoothly but partial over-reduction, i.e., partial hydrogenolysis to the corresponding phenol, occurred at higher H₂ pressures or upon longer exposures. The o-alkoxyphenol could be conveniently eliminated from the 1,2-dialkoxy-arene by basic workup (washing with aq NaOH and/or distillation over NaH).



Scheme 1. Preparation of bulky 1,2-dialkoxy-benzenes 1a-j.





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^{*} Corresponding authors. Fax: +33 148283011 (M.S); tel.: +386 14760250; fax: +386 14760300 (B.M); e-mail addresses: mstephan@phosphoenix.com (M. Stephan), barbara.mohar@ki.si (B. Mohar).

 $^{^\}dagger$ Present address: PhosPhoenix SARL, 115 rue de l'Abbé Groult, 75015 Paris, France.

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Table 1	
Bulky 1,2-dialkoxy-benzenes 1a – i prepared via Scheme 1	

1	\mathbb{R}^1	R ²	Conditions ^a	Isolated
				viold %
				yiciu //
a	<i>i</i> -Pr	Н	<i>i</i> -PrBr (1 equiv), K ₂ CO ₃ (2.5 equiv), reflux in acetone	88
b	<i>i</i> -Pr	<i>i</i> -Pr	<i>i</i> -PrI (3 equiv), K ₂ CO ₃ (3.5 equiv), reflux in acetone	94
с	<i>i</i> -Bu	Н	<i>i</i> -BuI (1.5 equiv), K ₂ CO ₃ (2.5 equiv), reflux in acetone	80
d	β-Met	β-Met	β-Methallyl chloride (3 equiv), KI (cat.), K ₂ CO ₃ (4 equiv), reflux in acetone	86
e ^b	<i>i</i> -Bu	<i>i</i> -Bu	H ₂ (1 bar), PtO ₂ (cat.), rt in EtOH	95
f	TMSCH ₂	TMSCH ₂	TMSCH ₂ I (2.2 equiv), K ₂ CO ₃ (4 equiv), 85 °C in DMF	85
g	Neopentyl	Н	<i>t</i> -BuCH ₂ I (2.2 equiv), K ₂ CO ₃ (4 equiv), 90 °C in DMF	67
h	c-Pen	c-Pen	c-PenBr or c-PenI (3 equiv), K ₂ CO ₃ (3.5 equiv), reflux in acetone	92
i	c-Hexen-2-yl	c-Hexen-2-yl	2-Cyclohexenyl bromide ^c (2.5 equiv), K ₂ CO ₃ (3 equiv), rt in acetone	86
j ^d	Су	Су	H ₂ (1 bar), PtO ₂ (cat.), rt in EtOH	86

^a Reactions were carried out with catechol using the appropriate alkyl halide under the mentioned conditions; the 1,2-dialkoxy-benzenes were distilled and displayed >99% chemical purity by ¹H NMR (see Experimental section).

^b Prepared from **1d**.

^c Another nomenclature is 3-bromocyclohexene.

^d Prepared from **1i**.

The nature of both alkyl chain (bulkiness) and halogen atom (especially iodine vs bromine) of the alkyl halide proved to dictate the proper choice of the reaction conditions, which in turn affected the reaction rate and yield. Whereas use of *i*-PrBr (1 equiv) or *i*-BuI (1-2 equiv) furnished mostly the corresponding *o*-alkoxyphenol, *i*-PrI (>2 equiv) afforded the targeted 1,2-diisopropoxybenzene in very high yield (after 2 day reflux in acetone). And while practically no O-alkylation of catechol in acetone in the presence of K₂CO₃ with the bulky neopentyl group occurred, using either its iodide or tosylate, mono-O-alkylation was complete in DMF at 90 °C using the iodide (even with>2 equiv) and 0,0'-dialkylation could only be achieved as described in the literature.^{4b} By contrast, under the same reaction conditions in DMF, bis(TMSCH₂)-alkylated product was obtained in good yield using TMSCH₂I. As expected, use of *i*-PrI and cyclopentyl iodide (c-PenI) in place of the often used corresponding bromides reduced significantly the reaction times. For example, using *i*-PrI led to cleaner and \sim twofold faster formation of **1b**.

We were also interested in the preparation of 1,2-diisopropoxynaphthalene (**1k**). Its synthesis was accomplished in a good overall yield (71%) starting from 2-isopropoxynaphthalene as depicted in Scheme 2. Noteworthy, the relatively sensitive formate intermediate ensuant from the Baeyer–Villiger oxidation of the aromatic carboxaldehyde was directly used in the subsequent transformation and its saponified product was O-alkylated in situ.



For example, bromoarenes **2a** and **2b** (Chart 1) were obtained in 56% and 64% isolated yield, respectively, following *o*-lithiation with *n*-BuLi/TMEDA of the corresponding arene and quench with 1,2-dibromoethane. Also, the contiguous bromine atom-containing 1,2-dialkoxy-arenes **2c**–**e** (Scheme 3) were prepared in good yields by etherification of the resulting *o*-bromo-hydroxyarene obtained by *ortho*-regioselective ring bromination (*t*-BuNH₂/Br₂⁹) of the *o*-alkoxyphenol.¹⁰ In particular, 3-bromocatechol acetonide (**2c**) was prepared in good overall yield starting from guaiacol. Thus, the 3-bromocatechol derived from 2-hydroxy-3-bromoanisole treatment with BBr₃, furnished **2c** (86% yield) under acid-catalyzed ketalization using PCl₃/acetone in benzene.¹¹ Such regioselective incorporation of bromine atom at an adjacent requisite position facilitates lithiation or prevents the subsequent lithiation in the 'wrong sense'.¹²



Scheme 2. Synthesis of 1,2-diisopropoxynaphthalene (1k).

2.2. Preparation of 1-bromo-2,3-dialkoxy-arenes

While 1,2-dialkoxy-arenes can be, in general, directly ring *ortho*lithiated with butyllithiums due to the *ortho*-directing ability of the alkoxy group,⁶ the generation of the *ortho*-lithiated species can be conveniently facilitated under mild conditions via



Chart 1. Prepared bromine-containing 1,2-dialkoxy-arenes.



Scheme 3. Synthesis of 2c-e.

2.3. Preparation of bulky isomeric trialkoxy-benzenes

The series of 1,2,3- (**3a**–**d**) and 1,3,5-trialkoxy-benzenes (**4a**,**d**) were prepared starting from pyrogallol and phloroglucinol, respectively, via Williamson etherification with the appropriate alkyl halide (Scheme 4, Table 2). Though pyrogallol is structurally more congested than phloroglucinol, here again various alkyl groups with branching on the first or second carbon atom were successfully incorporated in acetone. Interestingly, while monitoring periodically by ¹H NMR the pyrogallol O,O',O''-trialkylation progress, it was noticed that the unsymmetrical 2,3-dialkoxyphenol (not isolated) was the major intermediate formed and not the expected 2,6-dialkoxyphenol. Electronic effects seem to favor such a situation. In view of the clean formation of 1,2,3-triisopropoxy- (**3a**) and 1,2,3-tricyclopentoxybenzene (**3d**) in high yields, *i*-PrI and *c*-PenBr/*c*-PenI were used, respectively, however still 5–6-day reaction time was required.



Scheme 4. Preparation of bulky 1,2,3- (3a-d) and 1,3,5-trialkoxy-benzenes (4a,d).

 Table 2

 Bulky 1,2,3- (3a-d) and 1,3,5-trialkoxy-benzenes (4a,d) prepared via Scheme 4

	R	Conditions ^a	Isolated yield %
3a	<i>i</i> -Pr	<i>i</i> -PrI (4 equiv), K ₂ CO ₃ (5 equiv), reflux in acetone	91
3b	β-Met	β -Methallyl chloride (5 equiv), KI (cat.), K ₂ CO ₃	95
		(5 equiv), reflux in acetone	
3c ^b	i-Bu	H ₂ (1 bar), PtO ₂ (cat.), rt in EtOH	79
3d	c-Pen	<i>c</i> -PenBr (5 equiv), K ₂ CO ₃ (6 equiv), reflux in acetone ^c	76
4 a	i-Pr	<i>i</i> -PrBr (6 equiv), K ₂ CO ₃ (6 equiv), 60 °C in DMF	64
4d	c-Pen	<i>c</i> -PenBr (6 equiv), K ₂ CO ₃ (6 equiv), 60 °C in DMF	71

^a Reactions were carried out with pyrogallol or phloroglucinol using the appropriate alkyl halide under the mentioned conditions; the products were distilled and displayed >99% chemical purity by ¹H NMR (see Experimental section).

Prepared from 3b.

^c *c*-PenI (1 equiv) added after 3 days for higher conversion.

Notably, whereas three β -Met groups were successfully appended to pyrogallol using β -methallyl chloride, resulting in compound **3b** (within 2–3 days), the use of 2-cyclohexenyl bromide led to an unidentifiable complex mixture under various conditions. Also, deprotonation with NaH in THF at 0 °C followed by addition of 2-cyclohexenyl bromide did not improve the latter case in view of the targeted formation of 1,2,3-tri(2-cyclohexenyloxy)benzene. The greater steric demands, reactivity and sensitivity of the 2-cyclohexenyl group seem to prevent this access route. Gratifyingly, the preparation of the 1,2,3-tricyclohexyloxybenzene (**3f**) was cleanly achieved from 1,2-dicyclohexyloxybenzene (**1j**) by stepwise incorporation as outlined in Scheme 5. Thus, the recourse to activated allylic halides allowed the efficient preparation of **3b,c,e,f**. The PtO₂-catalyzed hydrogenation of the pendent olefinic groups in **3b,e** proceeded smoothly at 1 bar H₂, however operating at higher H₂ pressures or upon longer exposure times to H₂, partial hydrogenolysis to the phenol occurred as observed with the 1,2-dialkoxy-benzene counterparts.

While O,O',O"-trialkylation of phloroglucinol at 60 °C in DMF using *i*-PrBr afforded the desired compound **4a**,¹³ surprisingly no product was formed using *i*-PrI in refluxing acetone. This may be explained by the ability of phloroglucinol to form a keto tautomer under basic conditions.¹⁴

Further on, 1,2,3-tricyclohexyloxybenzene (**3f**) with its three contiguous and spacially close CyO groups was prepared in an indirect method starting from 1,2-dicyclohexyloxybenzene (**1j**) (Scheme 5). Thus, application of the O-alkylation—hydrogenation sequence on 2,3-dicyclohexyloxybenol furnished **3f** in 75% overall yield. The 1,2,4-tricyclohexyloxybenzene (**6**) isomer could be prepared from **1j** via formylation under Vilsmeier—Haack conditions (*N*-methylformanilide/POCl₃) followed by Baeyer—Villiger oxidation and saponification of the resulting formate, which furnished the 3,4-dicyclohexyloxyphenol. Its O-alkylation with 3-bromocyclohexene afforded **5** in 76% overall yield. Finally, hydrogenation of **5** under 1 bar H₂ provided **6**, while the thermal Claisen rearrangement of **5** led to the 2-(2-cyclohexenyl)-4,5-dicyclohexyloxyphenol.

2.3.1. Preparation of 1-bromo-2,3,4-trialkoxy-benzenes. As in the case of 1,2-dialkoxy-arenes, 1,2,3-trialkoxy-arenes can be directly o-lithiated with butyllithiums owing to the *ortho*-directing ability of the alkoxy group, though the latter require harsher conditions.^{2c,7a} Therefore, the generation of the *ortho*-lithiated species can be facilitated under mild conditions via the bromine—lithium exchange procedure of the suitably-placed bromine atom. The 1-bromo-2,3,4-trialkoxy-benzenes **7a,c,d,f** were conveniently prepared in high yields via an *ortho*-regioselective bromination procedure using NaBr/aq H₂O₂ in acetic acid¹⁵ (Scheme 6).

3. Conclusion

In summary, a large series of bulky 1,2-dialkoxy- and 1,2,3trialkoxy-benzenes was efficiently prepared. *ortho*-Regioselective incorporation of a bromine atom into these compounds was carried out, which would allow facile generation of the corresponding



Scheme 5. Synthesis of 1,2-dicyclohexyloxybenzene (1j) derivatives.



Scheme 6. Bromination of 1,2,3-trialkoxy-benzenes.

aryllithiums under mild conditions. Preparation of some pure isomeric 1,3,5- and 1,2,4-trialkoxy-benzenes was achieved as well. Their application in *P*-stereogenic phosphine synthesis will be communicated in due time.

4. Experimental section

4.1. General considerations

3-Bromocyclohexene (2-cyclohexenyl bromide)¹⁶ and 2,2dimethyl-2,3-dihydrobenzofuran-7-ol are commercially available. 2-Bromo-6-methoxyphenol and 2-bromo-6-isopropoxyphenol¹⁷ were prepared according to literature. ¹H (300 MHz, internal Me₄Si) and ¹³C (75 MHz, internal CDCl₃) NMR spectra were recorded for solutions in CDCl₃. IR spectra were recorded using a Specac Golden Gate ATR cell equipped with a diamond crystal.

4.2. Detailed synthetic protocols

4.2.1. 2-Isopropoxyphenol (**1a**). A mixture of catechol (11.0 g, 100 mmol), 2-bromopropane (9.40 mL, 100 mmol), and K₂CO₃ (34.5 g, 250 mmol) in acetone (100 mL) was refluxed for 1 day and the reaction progress monitored by TLC. The salts were filtered off rinsing with acetone and the filtrate concentrated. The residue was partitioned between diluted aq HCl (100 mL) and Et₂O (80 mL), and the product extracted with Et₂O (2×80 mL). The organic layer is dried (Na₂SO₄) and concentrated. Kugelrohr distillation (120 °C, 10 mbar) afforded a colorless liquid (13.4 g, 88%); $\delta_{\rm H}$ 1.35 (6H, d, *J* 6.1 Hz, 2Me), 4.56 (1H, sept, *J* 6.1 Hz, CH), 5.77 (1H, s, OH), 6.77–6.96 (4H, m). ¹H NMR is in accordance with the literature data.^{2b,17c}

4.2.2. 1,2-Diisopropoxybenzene (**1b**). A mixture of catechol (11.0 g, 100 mmol), 2-iodopropane (30.0 mL, 300 mmol), and K₂CO₃ (48.5 g, 351 mmol) in acetone (100 mL) was refluxed for 2 days and the reaction progress monitored by TLC. The salts were filtered off rinsing with acetone and the filtrate concentrated. The residue was partitioned between H₂O (100 mL) and Et₂O (80 mL), and the product extracted with Et₂O (2×80 mL). The organic layer was washed with 1 M NaOH (50 mL), dried (Na₂SO₄), and concentrated. Kugelrohr distillation (120 °C, 10 mbar) afforded a colorless liquid (18.3 g, 94%); *R*_f (toluene) 0.6; $\delta_{\rm H}$ 1.33 (12H, d, *J* 6.1 Hz, 4Me), 4.46 (2H, sept, *J* 6.1 Hz, 2CH), 6.87–6.96 (4H, m); $\delta_{\rm C}$ 22.2, 72.1, 118.4, 121.7, 149.2. ¹H NMR is in accordance with the literature data.¹⁸

4.2.3. 2-Isobutoxyphenol (1c). Prepared following similar procedure as for **1a** using isobutyl iodide (1-iodo-2-methylpropane) (17.3 mL, 150 mmol, 1.5 equiv) and K₂CO₃ (34.5 g, 250 mmol, 2.5 equiv) refluxing for 3 days. The residue was purified on silica gel eluting with hexane/toluene 95:5 affording a colorless liquid, which solidified upon standing (13.3 g, 80%); mp 50–51 °C; $\delta_{\rm H}$ 1.03 (6H, d, J 6.7 Hz, 2Me), 2.11 (1H, tsept, J 6.6, 6.7 Hz, CH₂CH), 3.79 (2H, d, J 6.6 Hz, CH₂), 5.67 (1H, s, OH), 6.81–6.88 (3H, m), 6.90–6.95 (1H, m); $\delta_{\rm C}$ 19.2, 28.2, 75.1, 111.6, 114.4, 120.0, 121.3, 145.8, 146.0; HRMS

(ESI): m/z calcd for $C_{10}H_{13}O_2$ [M^+ –H] 165.092, found 165.091. ¹H NMR is in accordance with the literature data.^{2a}

4.2.4. 1,2-Bis(β -methallyloxy)benzene (1d). Prepared following similar procedure as for **1b** using β -methallyl chloride (3-chloro-2-methylpropene) (27.2 g, 300 mmol, 3 equiv), KI (cat.), and K₂CO₃ (55.3 g, 400 mmol, 4 equiv) in acetone (200 mL) refluxing for 2 days. The residue was purified on silica gel eluting with petroleum ether 40–60/EtOAc 9:1 (R_f 0.7) affording a colorless liquid (18.8 g, 86%); ν_{max} 2915, 1592, 1500, 1451, 1247, 1204, 1123 cm⁻¹; δ_H 1.82–1.85 (6H, m, Me), 4.47–4.50 (4H, m, 2CH₂), 4.95–4.98 (2H, m, 2CH_aH_b), 5.10–5.12 (2H, m, 2CH_aH_b), 6.87–6.90 (4H, m); δ_C 19.3, 72.7, 112.3, 114.3, 121.2, 141.0, 148.8; m/z (ESI) 219.1 (100, MH⁺); HRMS (ESI): MH⁺, found 219.138. C₁₄H₁₉O₂ requires 219.139.

4.2.5. 1,2-Diisobutoxybenzene (**1e**). A mixture of **1d** (18.0 g, 82.5 mmol) and PtO₂ (40.0 mg) in EtOH (96%, 75 mL) was hydrogenated with a Parr hydrogenator at rt under 1 bar of H₂ for 2 h. Following, it was filtered through Celite and the filtrate concentrated. Kugelrohr distillation (115 °C, 0.04 mbar) over NaH afforded a faint yellow-colored oil (17.2 g, 95%); *R*_f (hexane/toluene 1:1) 0.6; ν_{max} 2957, 1592, 1501, 1470, 1452, 1251, 1224, 1121 cm⁻¹; $\delta_{\rm H}$ 1.04 (12H, d, *J* 6.7 Hz, 4Me), 2.12 (2H, tsept, *J* 6.6, 6.7 Hz, 2CH), 3.74 (4H, d, *J* 6.6 Hz, 2CH₂), 6.85–6.88 (4H, m); $\delta_{\rm C}$ 19.2, 28.4, 75.6, 114.1, 120.9, 149.4; *m*/*z* (ESI) 223.2 (100, MH⁺); HRMS (ESI): MH⁺, found 223.169. C₁₄H₂₃O₂ requires 223.170.

4.2.6. 1,2-Bis(trimethylsilylmethoxy)benzene (**1***f*). A mixture of catechol (1.10 g, 10.0 mmol), (iodomethyl)trimethylsilane (4.70 g, 22.0 mmol), and K₂CO₃ (5.50 g, 40.0 mmol) in DMF (40 mL) was stirred at 85 °C for 2 days. Following, H₂O (100 mL) was added and the product extracted with Et₂O (3×100 mL). The organic layer was washed with H₂O (4×50 mL), dried (Na₂SO₄), and concentrated affording a brown-colored oil, which was purified on silica gel eluting with hexane/CH₂Cl₂ 9:1 (R_f 0.4); colorless liquid (2.40 g, 85%); ν_{max} 2957, 1590, 1497, 1429, 1248, 1231, 1216, 1116 cm⁻¹; $\delta_{\rm H}$ 0.20 (18H, s, 9Me), 3.61 (4H, s, 2CH₂), 6.87–6.93 (2H, m), 6.96–7.02 (2H, m); $\delta_{\rm C}$ –3.1, 61.4, 112.0, 120.3, 151.5; *m*/*z* (ESI) 283.2 (100, MH⁺); HRMS (ESI): MH⁺, found 283.155. C₁₄H₂₇O₂Si₂ requires 283.155.

4.2.7. 2-Neopentoxyphenol (**1g**). Prepared following similar procedure as for **1f** using neopentyl iodide (1-iodo-2,2-dimethylpropane) (2.92 mL, 22.0 mmol, 2.2 equiv) and heating at 90 °C for 3 days. The resulting mixture was diluted with H₂O (50 mL), acidified and extracted with Et₂O (3×10 mL). The organic layer was washed with H₂O (3×10 mL), dried (Na₂SO₄), and concentrated affording a brown oil, which was purified on silica gel eluting with hexane/toluene 9:1 (R_f 0.2): yellowish syrup (1.21 g, 67%); $\delta_{\rm H}$ 1.06 (9H, s, 3Me), 3.68 (2H, s, CH₂), 5.62 (1H, s, OH), 6.81–6.89 (3H, m), 6.91–6.96 (1H, m); $\delta_{\rm C}$ 26.6, 31.9, 78.6, 111.7, 114.4, 120.1, 121.3, 145.8, 146.2; m/z (ESI) 181.1 (100, MH⁺); HRMS (ESI): MH⁺, found 181.122. C₁₁H₁₇O₂ requires 181.123.

4.2.8. 1,2-Dicyclopentoxybenzene (**1h**). Prepared following similar procedure as for **1b** using bromocyclopentane (29.0 mL, 270 mmol, 3 equiv) or iodocyclopentane (3 equiv). Kugelrohr distillation (90 °C, 0.04 mbar) over NaH afforded a colorless liquid (20.4–20.6 g, 92–93%); R_f (toluene) 0.7; v_{max} 2959, 2971, 1590, 1494, 1452, 1251, 1218, 1168 cm⁻¹; δ_H 1.53–1.66 (4H, m, 2CH₂), 1.75–1.96 (12H, m, 6CH₂), 4.71–4.77 (2H, m, 2 OCH), 6.84–6.92 (4H, m); δ_C 23.7, 32.6, 80.8, 116.9, 121.0, 148.8; m/z (EI) 246 (60, M⁺); HRMS (EI): M⁺, found 246.163. C₁₆H₂₂O₂ requires 246.162.

4.2.9. 1,2-Di(2-cyclohexenyloxy)benzene (**1i**). A mixture of catechol (22.0 g, 200 mmol), 2-cyclohexenyl bromide (80.5 g, 0.50 mol, 2.5 equiv), and K_2CO_3 (82.9 g, 0.60 mol, 3 equiv) in acetone

(250 mL) was stirred at rt for 1 day. The salts were filtered off rinsing with acetone and the filtrate concentrated. The residue was partitioned between H₂O (200 mL) and Et₂O (150 mL), and the product extracted with Et₂O (2×150 mL), dried (Na₂SO₄) and concentrated. Kugelrohr distillation (85 °C, 0.04 mbar) afforded a colorless liquid (46.5 g, 86%); *R*_f (hexane/EtOAc 9:1) 0.5; *v*_{max} 2934, 1590, 1492, 1451, 1247, 1205, 1110 cm⁻¹; $\delta_{\rm H}$ 1.60–2.17 (12H, m, 6CH₂), 4.69–4.75 (2H, m, 2 OCH), 5.85–5.96 (4H, m, 2CH=CH), 6.86–7.00 (4H, m); $\delta_{\rm C}$ 18.9, 19.0, 25.1, 28.49, 28.52, 73.1, 118.75, 118.78, 121.9, 127.0, 131.36, 131.38, 149.23, 149.24; *m/z* (EI) 270 (30, M⁺); HRMS (EI): M⁺, found 270.163. C₁₈H₂₂O₂ requires 270.162.

4.2.10. 1,2-Dicyclohexyloxybenzene (**1***j*). A mixture of 1,2-di(2-cyclohexenyloxy)benzene (**1***i*) (40.5 g, 150 mmol) and PtO₂ (100 mg) in EtOH (96%, 100 mL) was hydrogenated with a Parr hydrogenator at rt under 1 bar of H₂ for 6 h. Following, it was filtered through Celite and the filtrate concentrated. Kugelrohr distillation (130 °C, 0.04 mbar) over NaH afforded a colorless liquid (35.1 g, 86%), which crystallized upon standing; mp 40–42 °C; *Rf* (toluene/EtOAc 9:1) 0.4; *v*_{max} 2933, 1855, 1593, 1492, 1449, 1252, 1204, 1111 cm⁻¹; $\delta_{\rm H}$ 1.27–1.46 (6H, m, 3CH₂), 1.49–1.71 (6H, m, 3CH₂), 1.73–1.89 (4H, m, 2CH₂), 1.91–2.02 (4H, m, 2CH₂), 4.14–4.24 (2H, m, 2 OCH), 6.85–6.97 (4H, m); $\delta_{\rm C}$ 23.6, 25.6, 31.9, 77.0, 118.4, 121.5, 148.9; *m/z* (EI) 274 (10, M⁺); HRMS (EI): M⁺, found 274.194. C₁₈H₂₆O₂ requires 274.193.

4.2.11. 2-Isopropoxy-1-naphthaldehyde. To a cold (0 °C) mixture of 2-isopropoxynaphthalene^{19b} (10.0 g, 53.7 mmol) and *N*-methyl-formanilide (7.20 mL, 58.0 mmol) was added POCl₃ (5.30 mL, 58.0 mmol). Then it was heated at 75 °C for 12 h. The mixture was suspended in 2 M NaOH (90 mL) and extracted with Et₂O (3×100 mL). The organic layer was washed with 1 M HCl (50 mL), dried (Na₂SO₄), filtered through a bed of silica gel, and the filtrate concentrated. Crystallization from hexane afforded yellow crystals (11.3 g, 98%): $\delta_{\rm H}$ 1.42 (6H, d, *J* 6.0 Hz, 2Me), 4.79 (1H, sept, *J* 6.0 Hz, CH), 7.24 (1H, d, *J* 9.2 Hz), 7.39 (1H, ddd, *J* 1.1, 6.9, 7.9 Hz), 7.41 (1H, ddd, *J* 1.5, 7.0, 8.5 Hz), 7.70–7.77 (1H, m), 7.98 (1H, d, *J* 9.2 Hz), 9.25–9.30 (1H, m), 10.89 (1H, s); $\delta_{\rm C}$ 22.2, 72.4, 115.2, 118.0, 124.7, 124.8, 128.1, 128.4, 129.6, 131.6, 137.2, 162.9, 192.5. ¹H NMR is in accordance with the literature data.¹⁹

4.2.12. 2-Isopropoxy-1-naphthyl formate. To a cold (0 °C) solution of 2-isopropoxy-1-naphthaldehyde (10.0 g, 46.7 mmol) in CH₂Cl₂ (100 mL) was added under stirring *m*-CPBA (70%, 60.7 mmol, 15.0 g). The mixture was stirred at rt for 1 h then concentrated. The residue was triturated with hexane (70 mL), insolubles were filtered off, and the filtrate concentrated yielding the product as a yellow oil (10.2 g, 95%; contains <5% *m*-CBA); $\delta_{\rm H}$ 1.37 (6H, d, J 6.1 Hz, 2Me), 4.67 (1H, sept, J 6.1 Hz, CH), 7.32 (1H, d, J 9.1 Hz), 7.40 (1H, ddd, J 1.2, 6.8, 8.2 Hz), 7.51 (1H, ddd, J 1.3, 6.8, 8.4 Hz), 7.71–7.76 (1H, m), 7.78–7.82 (1H, m), 7.84–7.88 (1H, m), 8.42 (1H, s, CHO); $\delta_{\rm C}$ 22.2, 72.6, 116.9, 120.2, 124.5, 127.1 (2C), 127.7, 127.9, 129.3, 134.0, 145.8, 159.9; *m*/*z* (EI) 230 (40, M⁺); HRMS (EI): M⁺, found 230.094.

4.2.13. 1,2-Diisopropoxynaphthalene (**1k**). To a solution of 2isopropoxy-1-naphthyl formate (10.0 g, 43.4 mmol) in acetone/ MeOH 8:2 (100 mL) was added K₂CO₃ (12.0 g, 86.8 mmol). The mixture was stirred at rt for 1 h then 2-iodopropane (11.1 g, 65.1 mmol) was added and heated at 75 °C for 3 days. The concentrated residue was partitioned between Et₂O (3×100 mL) and H₂O (200 mL), and the organic layer dried (Na₂SO₄) then concentrated. Kugelrohr distillation (105 °C, 0.04 mbar) afforded a yellow oil (8.60 g, 81%); R_f (hexane/toluene 1:1) 0.6; ν_{max} 2974, 2931, 1594, 1467, 1378, 1371, 1262, 1237, 1106 cm⁻¹; δ_H 1.35 (12H, d, *J* 6.1 Hz, 4Me), 4.64 (1H, sept, *J* 6.1 Hz, OCH), 4.73 (1H, sept, *J* 6.1 Hz, OCH), 7.21 (1H, d, *J* 9.0 Hz), 7.33 (1H, ddd, *J* 1.3, 6.8, 8.2 Hz), 7.42 (1H, ddd, *J* 1.4, 6.8, 8.5 Hz), 7.50–7.54 (1H, m), 7.72–7.76 (1H, m), 8.13–8.18 (1H, m); $\delta_{\rm C}$ 22.5, 22.8, 72.3, 75.1, 119.6, 122.3, 123.4, 124.1, 125.5, 127.3, 130.1, 130.8, 142.8, 146.5; *m/z* (EI) 244 (15, M⁺); HRMS (EI): M⁺, found 244.147. C₁₆H₂₀O₂ requires 244.146.

4.2.14. 7-*Isopropoxy-2,2-dimethyl-2,3-dihydrobenzofuran*. Prepared following similar procedure as for **1b** from 2,2-dimethyl-2,3-dihydrobenzofuran-7-ol (16.4 g, 100 mmol) and 2-iodopropane (12.0 mL, 0.12 mol). Kugelrohr distillation (70 °C, 0.04 mbar) afforded a colorless liquid (19.2 g, 93%); R_f (toluene/EtOAc 9:1) 0.7; ν_{max} 2974, 1594, 1481, 1459, 1370, 1299, 1261, 1242, 1136, 1111 cm⁻¹; δ_H 1.34 (6H, d, *J* 6.1 Hz, CH*M*e₂), 1.49 (6H, s, CMe₂), 3.00 (2H, s, CH₂), 4.58 (1H, sept, *J* 6.1 Hz, OCH), 6.71–6.77 (3H, m); δ_C 22.0, 27.9, 43.0, 71.2, 86.5, 115.9, 117.5, 120.0, 128.3, 142.3, 148.5; m/z (ESI) 207 (60, MH⁺); HRMS (ESI): MH⁺, found 207.138. C₁₃H₁₉O₂ requires 207.139.

4.2.15. 6-Bromo-7-isopropoxy-2,2-dimethyl-2,3-dihydrobenzofuran a solution of 7-isopropoxy-2,2-dimethyl-2,3-(**2a**). To dihydrobenzofuran (0.62 g, 3.00 mmol) and TMEDA (0.45 mL, 3.00 mmol) in Et₂O (5 mL) was added under stirring *n*-BuLi (2 M in cyclohexane, 1.50 mL, 3.00 mmol). The resulting mixture was stirred at rt for 4 h, cooled to -40 °C and 1,2-dibromoethane (0.26 mL, 3.00 mmol) was added. After stirring at rt overnight, the reaction mixture was quenched with H₂O (1 mL). The residue was partitioned between H₂O (10 mL) and CH₂Cl₂ (10 mL), and the product extracted with CH₂Cl₂ (2×10 mL), dried (Na₂SO₄) and concentrated. Purification on silica gel eluting with hexane/toluene 7:3 afforded a colorless oil (0.48 g, 56%); *R*_f (hexane/toluene 1:1) 0.4; *v*_{max} 2975, 2930, 1589, 1448, 1371, 1313, 1101 cm⁻¹; $\delta_{\rm H}$ 1.32 (6H, d, / 6.2 Hz, CHMe₂), 1.47 (6H, s, CMe₂), 2.94 (2H, d, / 1.1 Hz, CH₂), 4.65 (1H, sept, / 6.2 Hz, OCH), 6.68 $(1H, td, 1.1, 7.9 Hz), 6.96 (1H, d, 17.9 Hz); \delta_{C} 22.4, 28.0, 42.8, 74.9, 88.1,$ 115.9, 120.0, 124.1, 128.5, 139.5, 151.5; *m*/*z*(ESI) 285 (25, MH⁺); HRMS (ESI): MH⁺, found 285.050. C₁₃H₁₈⁷⁹BrO₂ requires 285.049.

4.2.16. 3-Bromo-1,2-diisopropoxynaphthalene (**2b**). Prepared from 1,2-diisopropoxynaphthalene (**1k**) (4.88 g, 20.0 mmol) following similar procedure as for **2a**. Purification on silica gel eluting with hexane/toluene 8:2 afforded a colorless oil (4.14 g, 64%); *R*_f (hexane/toluene 1:1) 0.8; $\delta_{\rm H}$ 1.32 and 1.36 (12H, 2d, *J* 6.2 Hz, 4Me), 4.75 and 4.83 (2H, 2sept, *J* 6.2 Hz, 2CH), 7.37–7.49 (2H, m), 7.64–7.70 (1H, m), 7.84 (1H, s), 8.08–8.15 (1H, m); $\delta_{\rm C}$ 22.4, 22.6, 75.1, 75.6, 122.6, 125.77, 125.78, 126.5, 126.8, 130.2, 131.3, 133.6, 143.9, 146.2; *m*/*z* (EI) 322 (25, M⁺); HRMS (EI): M⁺, found 322.058. C₁₆H₁₉⁷⁹BrO₂ requires 322.057.

4.2.17. 1-Bromo-2,3-dihydroxybenzene (3-bromocatechol). To a cold (-78 °C) solution of 2-bromo-6-methoxyphenol (8.00 g, 39.4 mmol) in dry CH₂Cl₂ (40 mL) was added slowly BBr₃ (1 M in CH₂Cl₂, 43.1 mmol, 43.0 mL). After stirring at rt for 1 h, the mixture was poured onto ice-water, stirred for 30 min and extracted with CH₂Cl₂ (3×30 mL). The organic layer was dried (MgSO₄) then concentrated affording the title product as a brownish oil (6.90 g, 93%); R_f (toluene/EtOAc 9:1) 0.2; δ_H 5.60 (1H, br s, OH), 5.69 (1H, br s, OH), 6.72 (1H, dd, *J* 8.1, 8.1 Hz), 6.87 (1H, dd, *J* 1.5, 8.1 Hz), 7.00 (1H, dd, *J* 1.5, 8.1 Hz); δ_C 109.5, 114.9, 121.9, 123.3, 140.3, 144.5. ¹H NMR is in accordance with the literature data.²⁰

4.2.18. 3-Bromo-(isopropylidenedioxy)benzene (**2c**). To a cold (0 °C) solution of 3-bromocatechol (6.00 g, 31.7 mmol) and acetone (2.80 mL, 38.1 mmol) in benzene (25 mL) was added dropwise PCl₃ (1.11 mL, 12.7 mmol). The reaction mixture was stirred at rt until HCl evolution ceased, then H₂O (100 mL) was added. Extraction with CH₂Cl₂ (3×30 mL), drying (Na₂SO₄), and purification of the residue on silica gel eluting with toluene afforded a colorless oil (6.24 g, 86%); R_f (toluene) 0.5; δ_H 1.72 (6H, s, 2Me), 6.67 (2H, 2d, J 4.7 Hz), 6.91 (1H, dd, J 4.7, 4.7 Hz); δ_C 25.9, 100.7, 107.5, 118.7, 122.1,

124.3, 145.7, 147.7; m/z (ESI) 229.0 (100, MH⁺); HRMS (ESI): MH⁺, found 228.987. C₉H₁₀⁷⁹BrO₂ requires 228.986.

4.2.19. 1-Bromo-2-isopropoxy-3-methoxybenzene (2d). Prepared following similar procedure as for **1b** from 2-bromo-6-methoxyphenol (1.00 g, 4.92 mmol), 2-iodopropane (850 µL, 9.85 mmol, 2 equiv), and K₂CO₃ (1.36 g, 9.85 mmol). Colorless liquid (1.12 g, 93%); R_f (toluene) 0.6; ν_{max} 2976, 1572, 1473, 1450, 1435, 1259, 1232, 1102 cm⁻¹; δ_H 1.33 (6H, d, J 6.2 Hz, 2Me), 3.83 (3H, s, OMe), 4.55 (1H, sept, J 6.2 Hz, OCH), 6.83 (1H, dd, J 1.8, 8.3 Hz), 6.89 (1H, dd, J 7.7, 8.3 Hz), 7.13 (1H, dd, J 1.8, 7.7 Hz); δ_C 22.5, 560, 75.9, 111.6, 118.8, 124.3, 125.0, 144.7, 154.1.¹H NMR is in accordance with the literature data.^{17c}

4.2.20. 1-Bromo-3-isopropoxy-2-methoxybenzene (2e). Prepared following similar procedure as for **1b** from 2-bromo-6-isopropoxyphenol (6.70 g, 29.0 mmol), MeI (3.60 mL, 58.0 mmol, 2 equiv), and K₂CO₃ (8.00 g, 58.0 mmol, 2 equiv). Colorless liquid (6.7 g, 94%); R_f (hexane/EtOAc 9:1) 0.6; δ_H 1.36 (6H, d, J 6.1 Hz, 2Me), 3.86 (3H, s, OMe), 4.54 (1H, sept, J 6.1 Hz, OCH), 6.83–6.92 (2H, m), 7.12 (1H, dd, J 2.6, 6.9 Hz); δ_C 22.1, 60.4, 71.6, 115.4, 117.9, 124.7, 125.0, 147.8, 152.1; m/z (ESI) 267.0 (75, MNa⁺); HRMS (ESI): MNa⁺, found 266.999. C₁₀H₁₃⁷⁹BrO₂Na requires 267.000.

4.2.21. 1,2,3-Triisopropoxybenzene (**3a**). A mixture of pyrogallol (25.2 g, 0.20 mol), 2-iodopropane (80 mL, 0.80 mol), and K₂CO₃ (138 g, 1.00 mol) in acetone (300 mL) was refluxed for 5 days monitoring the reaction progress by TLC. The salts were filtered off rinsing with acetone and the filtrate concentrated. The residue was partitioned between H₂O (300 mL) and CH₂Cl₂ (100 mL), and the product extracted with CH₂Cl₂ (2×100 mL), dried (Na₂SO₄), and concentrated. Kugelrohr distillation (125 °C, 10 mbar) afforded a light yellow-colored liquid (45.9 g, 91%); *R*_f (toluene/EtOAc 9:1) 0.6; *v*_{max} 2975, 1584, 1468, 1382, 1372, 1244, 1109 cm⁻¹; $\delta_{\rm H}$ 1.30 (6H, d, J 6.2 Hz, 2Me), 1.32 (12H, d, J 6.1 Hz, 4Me), 4.33 (1H, sept, J 6.2 Hz, OCH), 4.52 (2H, sept, J 6.1 Hz, 2 OCH), 6.55 (2H, d, J 8.3 Hz), 6.88 (1H, t, J 8.3 Hz); $\delta_{\rm C}$ 21.8, 22.3, 70.7, 74.8, 109.3, 122.4, 139.2, 152.2; *m*/*z* (ESI) 253.2 (100, MH⁺); HRMS (ESI): MH⁺, found 253.180. C₁₅H₂₅O₃ requires 253.180.

4.2.22. 1,2,3-*Tris*(β-methallyloxy)benzene (**3b**). Prepared following similar procedure as for **3a** using β-methallyl chloride (3-chloro-2-methylpropene) (22.6 g, 250 mmol, 5 equiv), and KI (85.0 mg) refluxing for 2–3 days. Purification on silica gel eluting with hexane/CH₂Cl₂ 8:2 afforded a colorless oil (13.7 g, 95%); *R*_f (hexane/CH₂Cl₂ 7:3) 0.4; *v*_{max} 2916, 1598, 1490, 1473, 1451, 1249 cm⁻¹; $\delta_{\rm H}$ 1.82–1.85 (6H, m, 2Me), 1.88–1.91 (3H, m, Me), 4.43–4.46 (6H, m, 3CH₂), 4.90–4.93 (1H, m, CH), 4.94–4.98 (2H, m, 2CH), 5.08–5.12 (3H, m, 3CH), 6.56 (2H, d, *J* 8.3 Hz), 6.90 (1H, dd, *J* 8.0, 8.6 Hz); $\delta_{\rm C}$ 19.4, 19.7, 72.8, 76.9, 107.4, 112.4, 112.7, 123.2, 138.2, 141.0, 142.3, 153.0; *m*/*z* (ESI) 289.2 (100, MH⁺); HRMS (ESI): MH⁺, found 289.180. C₁₈H₂₅O₃ requires 289.180.

4.2.23. 1,2,3-*Triisobutoxybenzene* (**3***c*). A mixture of **3b** (12.0 g, 41.6 mmol) and PtO₂ (35.0 mg) in EtOH (96%, 50 mL) was hydrogenated with a Parr hydrogenator at rt under 1 bar of H₂ for 72 h. Following, it was filtered through Celite and the filtrate concentrated. Kugelrohr distillation (150 °C, 0.04 mbar) over NaH afforded a colorless liquid (9.68 g, 79%); R_f (hexane/toluene 2:8) 0.6; ν_{max} 2957, 1597, 1463, 1250 cm⁻¹; $\delta_{\rm H}$ 1.03 (12H, d, J 6.8 Hz, 4Me), 1.04 (6H, d, J 6.8 Hz, 2Me), 2.07 (1H, tsept, J 6.7, 6.7 Hz, CH), 2.11 (2H, tsept, J 6.7, 6.7 Hz, 2CH), 3.73 (4H, d, J 6.4 Hz, 2CH₂), 3.74 (2H, d, J 6.4 Hz, CH₂), 6.53 (2H, d, J 8.4 Hz), 6.90 (1H, t, J 8.4 Hz); $\delta_{\rm C}$ 19.3, 19.4, 28.5, 29.2, 75.4, 79.7, 106.6, 123.1, 138.3, 153.5; m/z (ESI) 295.2 (100, MH⁺); HRMS (ESI): MH⁺, found 295.226. C₁₈H₃₁O₃ requires 295.227.

4.2.24. 1,2,3-Tricyclopentoxybenzene (3d). Prepared following similar procedure as for 3a using bromocyclopentane (21.0 mL,

198 mmol, 5 equiv) refluxing for 6 days (with the addition of 1 equiv of iodocyclopentane after 3 days for higher conversion). Kugelrohr distillation (145 °C, 0.04 mbar) afforded yellowish crystals (10.0 g, 76%); R_f (toluene) 0.7; mp 30–32 °C; ν_{max} 2958, 1593, 1466, 1248, 1167 cm⁻¹; δ_H 1.50–1.69 (8H, m, 4CH₂), 1.74–1.98 (16H, m, 8CH₂), 4.66–4.72 (1H, m, CH), 4.72–4.79 (2H, m, 2CH), 6.51 (2H, d, *J* 8.2 Hz), 6.87 (1H, t, *J* 8.2 Hz); δ_C 23.4, 23.7, 32.66, 32.68, 79.9, 84.2, 108.0, 122.4, 138.4, 152.4; m/z (EI) 330 (15, M⁺); HRMS (EI): M⁺, found 330.220. C₂₁H₃₀O₃ requires 330.219.

4.2.25. 1,3,5-Triisopropoxybenzene (4a). A mixture of phloroglucinol dihydrate (11.35 g, 70.0 mmol), 2-bromopropane (40.0 mL, 0.42 mol, 6 equiv), and K₂CO₃ (58.0 g, 0.42 mol, 6 equiv) in DMF (150 mL) was heated at 60 °C for 1 day. Following, H₂O (100 mL) was added and the product was extracted with Et₂O (3×100 mL). The organic layer was washed with H₂O (4×50 mL), dried (Na₂SO₄), and concentrated. Purification on silica gel eluting with hexane/CH₂Cl₂ 9:1 followed by Kugelrohr distillation (90 °C, 0.04 mbar) afforded a colorless liquid (11.3 g, 64%); ν_{max} 2976, 1588, 1464, 1183, 1147, 1111 cm⁻¹; $\delta_{\rm H}$ 1.310 (18H, d, *J* 6.1 Hz, 6Me), 4.47 (3H, sept, *J* 6.1 Hz, 3OCH), 6.03 (3H, s); $\delta_{\rm C}$ 22.1, 69.7, 96.1, 159.7; *m/z* (ESI) 253.2 (100, MH⁺); HRMS (ESI): MH⁺, found 253.180. C₁₅H₂₅O₃ requires 253.180.

4.2.26. 1,3,5-*Tricyclopentoxybenzene* (**4d**). Prepared following similar procedure as for **4a** using bromocyclopentane (32.0 mL, 0.30 mol, 6 equiv) heating for 2 days. Purification on silica gel eluting with hexane/CH₂Cl₂ 9:1 followed by Kugelrohr distillation (190 °C, 0.04 mbar) afforded a yellowish liquid (11.7 g, 71%), which crystallized upon standing; mp 31–33 °C; v_{max} 2962, 1591, 1463, 1349, 1149 cm⁻¹; $\delta_{\rm H}$ 1.51–1.66 (6H, m, 3CH₂), 1.69–1.96 (18H, m, 9CH₂), 4.64–4.72 (3H, m, 3 OCH), 5.99 (3H, s); $\delta_{\rm C}$ 24.0, 32.9, 79.2, 95.5, 159.8; *m/z* (ESI) 331.2 (100, MH⁺); HRMS (ESI): MH⁺, found 331.228. C₂₁H₃₁O₃ requires 331.227.

4.2.27. 2,3-Dicyclohexyloxyphenol. To a solution of 1,2-dicyclohexyloxybenzene (1j) (10.0 g, 36.5 mmol) in cyclohexane (100 mL) was added under stirring t-BuLi (1.5 M in pentane, 29.2 mL, 43.8 mmol). The resulting mixture was heated at 60 °C for 4 h, cooled to -30 °C, and B(OMe)₃ (8.14 mL, 73.0 mmol) was slowly added. After stirring at rt for 15 min, the reaction mixture was hydrolyzed with 1 M HCl (200 mL), extracted with CH₂Cl₂ (3×50 mL), and dried (Na₂SO₄). The residue was taken in THF (250 mL) and 0.5 M NaOH (150 mL) was added followed by slow addition of 30% aq H_2O_2 (100 mL). The reaction mixture was stirred at rt overnight, cooled to 0 °C and Na₂S₂O₅ slowly added portionwise until evolution of oxygen ceased. This was acidified to pH ~ 2 with 1 M HCl and extracted with CH_2Cl_2 (3×50 mL), dried (Na₂SO₄), and concentrated. Purification on silica gel eluting with hexane/ toluene 6:4 afforded a yellowish oil (8.69 g, 82%); R_f (hexane/EtOAc 95:5) 0.4; $\delta_{\rm H}$ 1.13–1.64 (12H, m, 6CH₂), 1.69–1.85 (4H, m, 2CH₂), 1.90-2.09 (4H, m, 2CH₂), 4.20-4.31 (2H, m, 2 OCH), 5.87 (1H, br s, OH), 6.45 (1H, dd, / 1.4, 8.4 Hz), 6.55 (1H, dd, / 1.4, 8.2 Hz), 6.84 (1H, t, J 8.3 Hz); δ_C 23.5, 24.3, 25.4, 25.5, 31.8, 32.8, 75.9, 80.7, 107.1, 107.4, 123.2, 134.3, 150.3, 150.4; m/z (EI) 290 (50, M⁺); HRMS (EI): M⁺, found 290.190. C₁₈H₂₆O₃ requires 290.188.

4.2.28. 2,3-Dicyclohexyloxy-(2-cyclohexenyloxy)benzene (**3e**). Prepared following similar procedure as for **1i** from 2,3-dicyclohexyloxyphenol (8.50 g, 29.3 mmol), 3-bromocyclohexene (7.10 g, 43.9 mmol, 1.5 equiv), and K₂CO₃ (8.10 g, 58.6 mmol, 2 equiv) stirring at rt for 3 days. Kugelrohr distillation (60 °C, 0.04 mbar) afforded a yellow-colored syrup (10.1 g, 94%); R_f (hexane/EtOAc 95:5) 0.4; ν_{max} 2932, 2856, 1583, 1466, 1239 cm⁻¹; δ_H 1.15–1.41 (6H, m, 3CH₂), 1.45–2.17 (20H, m, 10CH₂), 4.00–4.10 (1H, m, OCH), 4.17–4.27 (1H, m, OCH), 4.73–4.79 (1H, m, OCH), 5.88–5.92 (2H, m, 2CH), 6.54–6.61 (2H, m), 6.86 (1H, t, J 8.2 Hz); δ_C

19.0, 23.8, 24.16, 24.19, 25.1, 25.6, 25.7, 28.5, 31.98, 32.01, 32.69, 32.73, 72.3, 76.6, 80.7, 109.9, 110.0, 122.5, 126.9, 133.4, 139.6, 152.4, 152.5; m/z (EI) 370 (55, M⁺); HRMS (EI): M⁺, found 370.252. C₂₄H₃₄O₃ requires 370.251.

4.2.29. 1,2,3-Tricyclohexyloxybenzene (**3f**). A mixture of **3e** (10.0 g, 27.0 mmol) and PtO₂ (50.0 mg) in EtOH (96%, 50 mL) was hydrogenated with a Parr hydrogenator at rt under 1 bar of H₂ for 1 h. Following, it was filtered through Celite and the filtrate concentrated affording a colorless oil (9.75 g, 97%): ν_{max} 2931, 2855, 1497, 1210, 1166 cm⁻¹; $\delta_{\rm H}$ 1.18–1.65 (18H, m, 9CH₂), 1.72–1.87 (6H, m, 3CH₂), 1.91–2.08 (6H, m, 3CH₂), 4.00–4.11 (1H, m, OCH), 4.17–4.27 (2H, m, 2 OCH), 6.54 (2H, d, J 8.3 Hz), 6.84 (1H, t, J 8.3 Hz); $\delta_{\rm C}$ 23.7, 24.2, 25.6, 25.7, 32.0, 32.7, 76.5, 80.6, 109.5, 122.4, 139.4, 152.3; *m*/*z* (EI) 372 (25, M⁺); HRMS (EI): M⁺, found 372.268. C₂₄H₃₆O₃ requires 372.266.

4.2.30. 3,4-Dicyclohexyloxybenzaldehyde. To a cold (0 °C) solution of **1j** (14.6 g, 53.4 mmol) in NMFA (8.66 g, 64.1 mmol, 7.91 mL) was added POCl₃ (9.83 g, 64.1 mmol, 5.87 mL). After overnight stirring at 70 °C, the reaction was quenched with 2 M NaOH (100 mL) and the product extracted with Et₂O (3×50 mL). The organic layer was dried (Na₂SO₄) and concentrated affording a yellow oil (15.5 g, 96%); R_f (toluene) 0.4; δ_H 1.28–1.47 (6H, m, 3CH₂), 1.49–1.71 (6H, m, 3CH₂), 1.74–1.89 (4H, m, 2CH₂), 1.90–2.03 (4H, m, 2CH₂), 4.23–4.33 (1H, m, OCH), 4.34–4.44 (1H, m, OCH), 6.99 (1H, d, J 8.1 Hz), 7.42 (1H, dd, J 2.0, 8.1 Hz), 7.44 (1H, d, J 2.0 Hz), 9.82 (1H, s, CHO); δ_c 23.3, 23.4, 25.4, 25.5, 31.4, 31.6, 76.5, 77.2, 114.9, 116.3, 126.2, 129.8, 148.7, 154.8, 190.7; m/z (EI) 302 (20, M⁺); HRMS (EI): M⁺, found 302.189. C₁₉H₂₆O₃ requires 302.188.

4.2.31. 3,4-Dicyclohexyloxyphenyl formate. To a cold (0 °C) solution of 3,4-dicyclohexyloxybenzaldehyde (15.0 g, 49.6 mmol) in CH₂Cl₂ (100 mL) was added under stirring *m*-CPBA (70%, 64.5 mmol, 15.9 g). The mixture was stirred at rt for 1 h then concentrated. The residue was triturated with hexane (50 mL), insolubles were filtered off, and the filtrate concentrated yielding the product as a yellow oil (14.4 g, 91%); R_f (hexane/EtOAc 9:1) 0.7; ν_{max} 2933, 2857, 1740, 1498, 1210, 1158 cm⁻¹; δ_H 1.24–1.42 (6H, m, 3CH₂), 1.48–1.65 (6H, m, 3CH₂), 1.73–1.87 (4H, m, 2CH₂), 1.89–2.01 (4H, m, 2CH₂), 4.09–4.25 (2H, m, 2 OCH), 6.64 (1H, dd, *J* 2.6, 8.7 Hz), 6.70 (1H, d, *J* 2.6 Hz), 6.91 (1H, d, *J* 8.7 Hz), 8.28 (1H, s, CHO); δ_C 23.6, 23.7, 25.5, 25.6, 31.7, 32.0, 77.2, 78.1, 110.5, 113.1, 119.1, 144.2, 146.9, 149.8, 159.7; *m/z* (ESI) 319.2 (40, MH⁺); HRMS (ESI): MH⁺, found 319.191. C₁₉H₂₇O₄ requires 319.191.

4.2.32. 3,4-Dicyclohexyloxyphenol. To a solution of 3,4-dicyclohexyloxyphenyl formate (14.0 g, 44.0 mmol) in MeOH (100 mL) was added under stirring K₂CO₃ (12.2 g, 88.0 mmol). The mixture was stirred at rt for 1 h, concentrated, and the residue partitioned between H₂O (200 mL) and CH₂Cl₂ (100 mL), and the product extracted with CH₂Cl₂ (2×100 mL). The organic layer was dried (Na₂SO₄), concentrated, and the residue crystallized from hexane to afford a white powder (12.1 g, 95%); R_f (hexane/EtOAc 9:1) 0.3; mp 96–98 °C; v_{max} 3455, 2935, 2857, 1605, 1508, 1449, 1203, 1168 cm⁻¹; δ_H 1.19–1.37 (6H, m, 3CH₂), 1.42–1.59 (6H, m, 3CH₂), 1.70-1.83 (4H, m, 2CH₂), 1.87-2.01 (4H, m, 2CH₂), 3.94-4.04 (1H, m, OCH), 4.13–4.23 (1H, m, OCH), 5.88 (1H, br s, OH), 6.29 (1H, dd, J 2.9, 8.5 Hz), 6.43 (1H, d, J 2.9 Hz), 6.77 (1H, d, J 8.5 Hz); δ_C 23.7, 24.0, 25.58, 25.63, 31.8, 32.2, 76.7, 79.4, 104.8, 107.0, 121.4, 141.7, 150.3, 151.4; *m*/*z* (ESI) 291.2 (100, MH⁺); HRMS (ESI): MH⁺, found 291.195. C₁₈H₂₇O₃ requires 291.196.

4.2.33. 3,4-Dicyclohexyloxy-(2-cyclohexenyloxy)benzene (5). Prepared from 3,4-dicyclohexyloxyphenol (10.0 g, 34.4 mmol) following similar procedure as for **3e** and stirring at rt for 2–3 days. Colorless syrup (11.7 g, 92%); R_f (hexane/EtOAc 7:3) 0.9; ν_{max} 2932, 2855, 1511, 1420, 1182, 1167 cm⁻¹; δ_H 1.21–1.40 (6H, m, 3CH₂), 1.44–1.67 (7H, m, 3CH₂, CH), 1.70–2.19 (13H, m, 6CH₂, CH), 3.96–4.06 (1H, m, OCH), 4.14–4.24 (1H, m, OCH), 4.64–4.71 (1H, m, OCH=CH), 5.82–5.89 (1H, m, CH=CH), 5.91–5.99 (1H, m, CH=CH), 6.41 (1H, dd, J 2.8, 8.7 Hz), 6.54 (1H, d, J 2.8 Hz), 6.83 (1H, d, J 8.7 Hz); δ_C 18.9, 23.6, 23.8, 25.0, 25.6, 25.7, 28.2, 31.83, 31.84, 32.2, 71.2, 76.6, 78.7, 106.6, 106.9, 120.7, 126.5, 131.8, 142.5, 150.3, 151.1; m/z (ESI) 371.3 (100, MH⁺); HRMS (ESI): MH⁺, found 371.259.

4.2.34. 1,2,4-Tricyclohexyloxybenzene (**6**). Prepared from **5** (10.0 g, 27.0 mmol) following similar procedure as for **3f**. Colorless oil (9.65 g, 96%); R_f (toluene) 0.7; v_{max} 2931, 2855, 1497, 1210, 1167 cm⁻¹; δ_H 1.22–1.64 (18H, m, 9CH₂), 1.73–1.86 (6H, m, 3CH₂), 1.90–2.03 (6H, m, 3CH₂), 3.96–4.06 (1H, m, OCH), 4.05–4.15 (1H, m, OCH), 4.15–4.25 (1H, m, OCH), 6.39 (1H, dd, *J* 2.9, 8.8 Hz), 6.51 (1H, d, *J* 2.9 Hz), 6.82 (1H, d, *J* 8.8 Hz); δ_C 23.8, 23.9, 24.0, 25.6, 25.67, 25.7, 31.97, 31.99, 32.3, 76.1, 76.9, 78.8, 107.1, 107.5, 120.7, 142.6, 150.3, 153.2; m/z (ESI) 373.3 (40, MH⁺); HRMS (ESI): MH⁺, found 373.274. C₂₄H₃₇O₃ requires 373.274.

4.2.35. 2-(2-Cyclohexenyl)-4,5-dicyclohexyloxyphenol. Heating **5** (1.00 g, 2.70 mmol) at 150 °C for 1–2 h afforded the pure title compound (0.75 g, 75%); R_f (toluene/EtOAc 95:5) 0.5; mp 150–152 °C; δ_H 1.20–1.39 (6H, m, 3CH₂), 1.42–1.69 (8H, m, 4CH₂), 1.70–1.85 (5H, m, CH, 2CH₂), 1.87–2.01 (5H, m, CH, 2CH₂), 2.06–2.14 (2H, m, CH₂), 3.34–3.46 (1H, m, CH), 3.92–4.05 (1H, m, OCH), 4.08–4.22 (1H, m, OCH), 5.33 (1H, br s, OH), 5.78 (1H, m, CH), 6.00 (1H, m, CH), 6.41 (1H, s), 6.68 (1H, s); δ_C 21.3, 23.7, 24.0, 25.0, 25.6, 25.7, 30.2, 31.9, 32.3, 37.4, 76.8, 79.1, 105.7, 121.9, 122.9, 130.0, 130.6, 141.6, 148.6, 149.1; *m*/z (ESI) 371.3 (100, MH⁺); HRMS (ESI): MH⁺, found 371.259. C₂₄H₃₅O₃ requires 371.259.

4.2.36. 1-Bromo-2,3,4-triisopropoxybenzene (**7a**). To a mixture of **3a** (5.00 g, 19.8 mmol), NaBr (2.00 g, 19.8 mmol) in acetic acid (20 mL)/CH₂Cl₂ (20 mL) was added dropwise 30% aq H₂O₂ (2.00 mL) at rt and the reaction progress was followed by TLC. After 2 h, H₂O (30 mL) was added and the product extracted with CH₂Cl₂ (3×50 mL). The organic layer was dried (Na₂SO₄), concentrated and the residue purified on silica gel eluting with hexane/toluene 3:2 to afford a colorless liquid (6.23 g, 95%); *R*_f (hexane/toluene 1:1) 0.5; ν_{max} 2976, 1462, 1434, 1382, 1372, 1103 cm⁻¹; $\delta_{\rm H}$ 1.28 (6H, d, *J* 6.1 Hz, 2Me), 1.31 (6H, d, *J* 6.1 Hz, 2Me), 1.34 (6H, d, *J* 6.1 Hz, 2Me), 4.42 (1H, sept, *J* 6.1 Hz, OCH), 6.54 (1H, d, *J* 9.0 Hz), 7.15 (1H, d, *J* 9.0 Hz); $\delta_{\rm C}$ 2.0, 224, 22.5, 70.9, 75.5, 75.7, 109.7, 111.1, 126.4, 142.8, 149.8, 152.1; *m/z* (EI) 330 (25, M⁺); HRMS (EI): M⁺, found 330.084. C₁₅H₂₃⁷⁹BrO₃ requires 330.083.

4.2.37. 1-Bromo-2,3,4-triisobutoxybenzene (**7c**). Prepared from **3c** (5.89 g, 20.0 mmol) following similar procedure as for **7a**. Colorless oil (6.94 g, 93%); R_f (hexane/CH₂Cl₂ 85:15) 0.5; δ_H 1.03 (6H, d, *J* 6.8 Hz, 2Me), 1.04 (6H, d, *J* 6.8 Hz, 2Me), 1.06 (6H, d, *J* 6.8 Hz, 2Me), 2.07 (1H, tsept, *J* 6.5, 6.8 Hz, CH), 2.12 (1H, tsept, *J* 6.5, 6.8 Hz, CH), 2.13 (1H, tsept, *J* 6.5, 6.8 Hz, CH), 3.69 (2H, d, *J* 6.5 Hz, CH₂), 3.76 (2H, d, *J* 6.5 Hz, CH₂), 3.81 (2H, d, *J* 6.5 Hz, CH₂), 6.53 (1H, d, *J* 9.0 Hz); δ_C 19.28, 19.31, 19.4, 28.4, 29.17, 29.20, 75.3, 80.22, 80.24, 108.5, 109.3, 126.5, 143.2, 150.6, 153.1; *m*/z (ESI) 273.1 (30, MH⁺); HRMS (ESI): MH⁺, found 373.137. C₁₈H₃₀⁷⁹BrO₃ requires 373.138.

4.2.38. 1-Bromo-2,3,4-tricyclopentoxybenzene (**7d**). Prepared from **3d** (6.61 g, 20.0 mmol) following similar procedure as for **7a**. Colorless liquid (7.29 g, 89%); R_f (hexane/toluene 1:1) 0.7; ν_{max} 2959, 1463, 1432, 1289, 1208 cm⁻¹; δ_H 1.51–1.73 (10H, m, 5CH₂), 1.76–1.99

(14H, m, 7CH₂), 4.67–4.74 (1H, m, OCH), 4.74–4.80 (1H, m, OCH), 5.02–5.08 (1H, m, OCH), 6.51 (1H, d, J 9.0 Hz), 7.14 (1H, d, J 9.0 Hz); $\delta_{\rm C}$ 23.6, 23.7, 24.0, 32.7, 32.8, 80.0, 84.8, 85.2, 109.2, 110.2, 126.3, 142.3, 149.7, 152.3; *m/z* (EI) 408 (10, M⁺); HRMS (EI): M⁺, found 408.131. C₂₁H₂₉⁷⁹BrO₃ requires 408.130.

4.2.39. 1-Bromo-2,3,4-tricyclohexyloxybenzene (**7f**). Prepared from **3f** (7.45 g, 20.0 mmol) following similar procedure as for **7a**. Colorless oil (8.03 g, 89%); *R*_f (hexane/EtOAc 95:5) 0.6; ν_{max} 2932, 2856, 1462, 1449, 1434, 1206 cm⁻¹; $\delta_{\rm H}$ 1.16–1.65 (m, 18H; 9CH₂), 1.71–1.87 (6H, m, 3CH₂), 1.90–2.06 (6H, m, 3CH₂), 4.05–4.15 (1H, m, OCH), 4.15–4.25 (1H, m, OCH), 4.31–4.42 (1H, m, OCH), 6.53 (1H, d, *J* 9.3 Hz), 7.13 (1H, d, *J* 9.3 Hz); $\delta_{\rm C}$ 23.8, 24.3, 24.4, 25.56, 25.62, 31.8, 32.67, 32.7, 76.2, 81.1, 81.4, 109.4, 110.7, 126.2, 142.5, 149.6, 151.9; *m*/z (EI) 450 (45, M⁺); HRMS (EI): M⁺, found 450.178. C₂₄H₃₅⁷⁹BrO₃ requires 450.177.

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References and notes

- For natural occurring compounds containing them, see: (a) Studies in Natural Product Chemistry, Bioactive Natural Products (Part O); Atta-ur-Rahman, Ed.; Elsevier Ltd: Amsterdam, 2008; (b) For examples of their use in material chemistry, see: Freudenmann, R; Behnisch, B.; Hanack, M. J. Mater. Chem. 2001, 11, 1618–1624; (c) Yelamaggad, C. V.; Achalkumar, A. S.; Rao, D. S. S.; Prasad, S. K. J. Org. Chem. 2007, 77, 8308–8318.
- For a particularly interesting strategy to *o*-alkoxyphenols from salicylaldehyde, see: (a) Syper, L. Synthesis **1989**, 3, 167–172; (b) Ishikawa, T.; Mizutani, A.; Miwa, C.; Oku, Y.; Komano, N.; Takami, A.; Watanabe, T. *Heterocycles* **1997**, 45, 2261–2272; (c) For an example of preparation of some 1,2-dialkoxy- and 1,2,3trialkoxy-benzenes, see: Horner, L.; Simons, G. *Phosphorus and Sulfur* **1983**, *14*, 189–209.
- (a) Zupančič, B.; Mohar, B.; Stephan, M. Adv. Synth. Catal. 2008, 350, 2024–2032; (b) Zupančič, B.; Mohar, B.; Stephan, M. Tetrahedron Lett. 2009, 50, 7382–7384.

- 4. (a) Our results regarding attempts to prepare 1,2-di-*tert*-butoxybenzene will be published as a separate full study as they are in disagreement with the reported ones in Ref. 2c. Therein, also the preparation of **1b** (48% yield) and **1e** (42% yield) has been reported using the corresponding alkyl bromides refluxing in acetone for 8 days; (b) Preparation of 1,2-dineopentoxybenzene was achieved using neopentyl tosylate (>2 equiv) and KOH under phase transfer catalysis in HMPA. For this, see: Khanamiryan, A. K.; Bhardwaj, N.; Leznoff, C. C. J. Porphyrins Phthalocyanines **2000**, *4*, 484–490.
- For other particular procedures for preparing o-(2-cyclohexyloxy)- or o-(2-cyclohexenyloxy)phenol, see: (a) Melis, S.; Piras, P. P.; Plumitallo, A.; Sotgiu, F. J. Heterocycl. Chem. 1983, 20, 1413–1414; (b) Tietze, L. F.; Lohmann, J. K.; Stadler, C. Synlett 2004, 1113–1116.
- (a) Wakefield, B. J. Organolithium Methods; Academic: London, 1990; (b) Clayden, J. Organolithiums: Selectivity for Synthesis; Elsevier Science Ltd: Oxford, 2002.
- For effects of MeO groups on benzene metallation, see: (a) Maggi, R.; Schlosser, M. *Tetrahedron Lett.* **1999**, *40*, 8797–8800; (b) The low-yield direct o-lithiation of (isopropylidenedioxy)benzene was not successful in our hands as described in the literature. For the latter, see: Huang, D.-S.; Ting, S.-H. *J. Chem. Res., Synop.* **1994**, *12*, 500–501.
- The radical halogenation of 1,2-dialkoxy-arenes leads to the corresponding 4halo derivatives. For an example, see: James, C. A.; Snieckus, V. J. Org. Chem. 2009, 74, 4080–4093.
- 9. For ortho-bromination of phenols, see: Pearson, D. E.; Wysong, R. D.; Breder, C. V. J. Org. Chem. **1967**, 32, 2358–2360.
- 10. Interestingly, 2-bromo-6-trimethylsilylanisole was prepared by selective transmetallation of 2,6-dibromophenol followed by reaction with TMSCI and etherification with MeI. For this, see Ref. 3a (Supporting Information).
- For a ketalization of catechols with acetone, see: Ivanov, A. V.; Svinareva, P. A.; Tomilova, L. G.; Zefirov, N. S. Russ. Chem. Bull., Int. Ed. 2001, 50, 919–920.
- 12. For an example of bromine-lithium exchange using 2e, see Ref. 3b.
- 13. Chaumeil, H.; Signorella, S.; Le Drian, C. *Tetrahedron* **2000**, *56*, 9655–9662 and personal communication from the corresponding author.
- Note, 1,3-diisopropoxybenzene could be prepared in >95% yield under standard Williamson etherification of resorcinol using *i*-Prl in acetone. For a study of preparation of phloroglucinol linear ethers, see: Touchstone, J. C.; Ashmore, J.; Huffan, M. N. *J. Am. Chem. Soc.* **1956**, 78, 5643–5645.
- 15. Yang, J.; Weng, L.; Zheng, H. Synth. Commun. 2006, 2401-2405.
- This reagent can be prepared as described in: Ponomarev, N. E.; Stambirskii, M. V.; Dvorko, G. F.; Bazil'chuk, A. V. Russ. J. Org. Chem. 2004, 40, 489–496.
- (a) Bungard, C. J.; Morris, J. C. Synthesis **2001**, *5*, 741–744; (b) Ishizaki, M.; Ozaki, K.; Kanematsu, A.; Isoda, T.; Hoshino, O. J. Org. Chem. **1993**, *58*, 3877–3885; (c) Kissau, L.; Stahl, P.; Mazitschek, R.; Giannis, A.; Waldmann, H. J. Med. Chem. **2003**, *46*, 2917–2931.
- 18. Buckley, T. F.; Rapoport, H. J. Am. Chem. Soc. 1980, 102, 3056-3062.
- (a) Pansare, S. V.; Ravi, R. G. Synlett **1994**, 823–824; (b) Barbasiewicz, M.; Szadkowska, A.; Makal, A.; Jarzembska, K.; Wozniak, K.; Grela, K. Chem.—Eur. J. **2008**, *14*, 9330–9337 (we used acetone instead of MeCN for the preparation of 2-isopropoxynaphthalene).
- 20. Hansen, T. V.; Skattebøl, L. Tetrahedron Lett. 2005, 46, 3357-3358.